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⑯ **Latex based adhesive prepared by emulsion polymerization.**

⑯ A latex based adhesive characterized by a first component (adhesive component) comprising an interpolymer of at least one hard monomer, at least one soft monomer, and optionally a copolymerizable carboxylic acid comonomer, having a Tg of from -70°C to -5°C and a second polymeric component (tackifier component) that is compatible with the adhesive component having a molecular weight less than 5,000 and a Tg from -10°C to 50°C, both the adhesive component and tackifier component having been separately prepared by emulsion polymerization techniques.

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The present invention relates to an adhesive. More particularly, the present invention relates to such an adhesive having one polymeric component adapted to provide adhesive properties and another polymeric component adapted to provide tackifying properties. More particularly still, the present invention relates to such an adhesive wherein both the adhesive and tackifier components are separately prepared by emulsion polymerization.

Aqueous colloidal dispersions of polymers, which are referred to in the art as latexes, are generally known to be useful as adhesives. A wide variety of adhesive formulations have been prepared that are specially adapted for particular uses. While latexes generally provide good adhesion to various substrates, it is often desirable to include a secondary component, generally a low molecular weight emulsified resin which is employed in order to provide initial stickiness or tack to the adhesive formulation. Such a component is referred to as a tackifier. Certain prior art adhesive formulations have included a latex based adhesive component such as a latex of a styrene/butadiene copolymer, and conventional tackifiers such as rosin esters, or other suitable tackifier, see e.g. U.S. Patent 4,189,419. The properties of the adhesive composition may be adjusted by appropriate selection of both the type and amounts of tackifier or adhesive component employed in order to provide a resulting composition having the desired properties.

However, such adhesive formulations have certain undesirable properties. For example, the process for dispersing the tackifying resin adds unnecessary complexity and additional cost to the formulation. In addition, such formulations often demonstrate insufficient coalescence between resin particles and latex particles, thereby necessitating using excessive amounts of tackifier resin. This problem has been previously disclosed in U.S. Patent 4,445,959, Column 12. Also, because relatively high levels of surfactant are required to disperse the tackifying resin adhesive properties are often detrimentally affected. Finally, under some conditions of high shear such as may be encountered during application of adhesives, these previously known adhesive formulations may lack sufficient stability.

Techniques are already known for the preparation of low molecular weight polymers via free-radical, emulsion polymerization processes. Typically, such techniques include initiation by organic peroxides or inorganic persulfate compounds, and the use of large quantities of a suitable chain transfer agent such as a halogen compound, or a mercaptan compound.

Latex based adhesive components, especially copolymers of styrene and butadiene are similarly known. Suitable examples of such adhesive components are disclosed in U.S. Patent 4,419,481.

In GB Patent 2,137,212, there is disclosed a tackified styrene/butadiene polymer adhesive made by preparing an emulsion polymerized adhesive component in the presence of an emulsion of a synthetic tackifier resin or rosin derivative.

In U.S. Patent 4,438,232, a latex adhesive resin is disclosed. In U.S. Patents 4,013,824, 4,064,377 and 4,145,494, emulsion processes for preparing polymers with a mercaptan as a chain transfer agent are disclosed.

It would be desirable to prepare by emulsion polymerization a latex of an ultra low molecular weight polymer suitable for use as a tackifying resin which does not require dispersing a solid polymer in water by use of surfactant.

In addition, it would be desirable to provide a latex based adhesive formulation which does not require the presence of a conventional, emulsified tackifier resin.

Finally, it would be desirable to provide an adhesive formulation wherein both the adhesive component and the tackifier component are prepared entirely by emulsion polymerization.

According to the present invention, there is now provided a latex based adhesive, the adhesive characterized by having at least two polymeric components, the first polymeric component (adhesive component) comprising a random interpolymer of at least one hard monomer and at least one soft monomer and optionally a copolymerizable carboxylic acid and having a glass transition temperature, Tg, of from -70°C to -5°C, and the second polymeric component (tackifier component) that is compatible with the adhesive component, and having a number average molecular weight less than 5,000 and a Tg from -10°C to 50°C, provided further that both the adhesive component and tackifier component have been separately prepared by emulsion polymerization techniques.

The latex based adhesive of the invention is particularly useful as a pressure sensitive adhesive for coating conventional substrates to form adhesive tapes in the preparation of carpet backings. It may also be employed as a laminating adhesive, such as in film to film laminations or film to paper laminations and in other adhesive applications.

Desirably, the adhesive component comprises from 95 to 20 percent and the tackifier component comprises from 5 to 80 percent of the invented composition, measured on a dry weight basis.

As used herein, "glass transition temperature" (Tg) is determined by Dynamic Mechanical Spectroscopy (DMS) or by Differential Scanning Calorimetry (DSC) at a rate of 10°C/min. utilizing a duPont 1090

model or equivalent calorimeter. Moreover, as used herein "compatibility" may be said to exist if respective polymers, upon mixing, demonstrate a single Tg or a narrowing of the difference in separately determined Tg's.

Molecular weights of polymers are measured by size exclusion chromatography (SEC) using a 2-4000 5 HXL and 1-2500 HXL column calibrated for low molecular weight polymers. Samples are prepared utilizing approximately 0.1 gram of polymer dissolved in 20 milliliters of tetrahydrofuran.

By the term "soft monomer" is meant addition polymerizable monomers which, if homopolymerized, would yield a homopolymer having a glass transition temperature (Tg) less than 25°C. Suitable examples of soft monomers include esters of unsaturated acids and saturated alcohols having from 3 to 15 carbons, 10 such as n-butyl acrylate, 2-ethylhexyl acrylate, cyclohexyl acrylate, and butyl methacrylate; and conjugated aliphatic dienes having from 4 to 12 carbons such as 1,3-butadiene, 2-methyl-1,3-butadiene, pentadiene, and 2-chloro-1,3-butadiene. Preferred soft monomers are n-butyl acrylate and 1,3-butadiene. A most preferred soft monomer is 1,3-butadiene.

By the term "hard monomers" is meant addition polymerization monomers other than carboxylic acids, 15 which if homopolymerized, would yield a homopolymer having a Tg greater than 25°C. Suitable examples of hard monomers include esters of unsaturated acids having from 3 to 10 carbons such as tert-butyl acrylate, methyl methacrylate, and tert-butyl methacrylate; and C₈₋₁₆ monovinylidene aromatic monomers such as styrene, α-methylstyrene, vinyl toluenes, bromostyrene, etc. A preferred hard monomer is styrene.

Suitable copolymerizable carboxylic acids are C₂₋₁₀ ethylenically unsaturated carboxylic acids such as 20 itaconic acid, acrylic acid, and maleic acid. A preferred copolymerizable carboxylic acid is itaconic acid.

Preferably, the tackifier component comprises an interpolymer of at least one hard monomer and optionally one or more soft monomers as well as a copolymerizable ethylenically unsaturated carboxylic acid. In a highly preferred embodiment, the tackifier component is prepared by emulsion polymerization of the soft and hard monomers, and optionally a carboxylic acid containing monomer in the presence of a 25 large quantity of chain transfer agent. By the term "large quantity" is meant a sufficient quantity of chain transfer agent to provide a latex of a interpolymer having a molecular weight of less than 5,000. As used herein, molecular weights are determined by the techniques of hydrodynamic chromatography. Suitable chain transfer agents include carbon tetrachloride, bromoform, and alkyl or aralkyl mercaptans having 2 to 20 carbons. Preferred chain transfer agents are aliphatic mercaptans containing from 8 to 20 carbon atoms. 30 A highly preferred aliphatic mercaptan is tertiary dodecyl mercaptan. Desirably, the chain transfer agent is employed in an amount from 1 percent to 20 percent by weight, preferably from 5 to 15 percent by weight based on reacted monomer weight.

Respective amounts of monomers are employed in the polymerization to prepare a tackifier component comprising from 30 percent to 100 percent by weight hard monomer, from 70 percent to 0 percent soft 35 monomer and up to 10 percent of a copolymerizable carboxylic acid. A preferred composition comprises from 60 to 90 percent hard monomer, 40 to 10 percent soft monomer, and 1 to 5 percent carboxylic acid. As used throughout the specification and claims, such values are based on total ethylenically unsaturated monomer content of the polymer latex and exclude polymerized mercaptan content. A most preferred tackifier component comprises an interpolymer of styrene, butadiene and an unsaturated carboxylic acid. 40 The number average molecular weight of the tackifier component is preferably from 1,000 to 4,000.

Suitable adhesive component interpolymers comprise from 30 to 90 percent soft monomer, from 70 to 10 percent hard monomer, and optionally up to 10 percent of a copolymerizable carboxylic acid comonomer. A preferred composition comprises from 50 to 80 percent soft monomer, from 50 to 20 percent hard monomer, and 1 to 5 percent unsaturated carboxylic acid. The foregoing percentages are 45 based on total ethylenically unsaturated monomer weight. The adhesive component is most preferably an interpolymer of styrene, butadiene and a copolymerizable carboxylic acid. A mercaptan, if present, is employed in an amount from 0.01 to 5.0 percent of the adhesive component.

The adhesive component may additionally comprise a crosslinking monomer such as divinyl benzene, trimethylol propane triacrylate, etc.

50 The adhesive and tackifier components of this invention may be prepared using conventional emulsion polymerization techniques. Thus, for example, the monomers and chain transfer agent employed are typically dispersed with agitation sufficient to emulsify the mixtures in an aqueous medium containing known emulsifying agents (i.e., surfactants) as well as other ingredients conventionally employed in the art as polymerization aids (e.g., chelating agents, seed latex, bactericides, antifoamants). Such monomers are then subjected to polymerization with the aid of a conventional source for generating free radicals, such as free radical initiators or activating radiation. The polymerization may be conducted in continuous or batch mode.

The final product is prepared by combining the respective components utilizing conventional com-

pounding or blending equipment. Additional conventional additives such as plasticizers, surfactants, fillers, antioxidants, pigments, thixotropic aids, and flow aids, may also be incorporated into the finished formulation.

Having described the invention, the following examples are provided as further illustrative of the invention and are not to be construed as limiting. Unless otherwise indicated, all parts, percentages, and ratios herein are based on dry solids weight.

Manufacture of the Tackifier Component

10 Sample 1

- (a) A monomer feed charge was prepared comprising 98 parts of styrene, 10 parts t-dodecyl mercaptan and 5 parts of carbon tetrachloride.
- (b) A second monomer feed charge was prepared comprising 18.7 parts of water, 2 parts of itaconic acid and 0.14 part of sodium hydroxide.
- (c) An aqueous feed charge was prepared comprising 22.0 parts of water, 0.5 parts sodium dodecyl benzene sulfonate surfactant, and 0.7 parts sodium persulfate.

To a one gallon (3.8 liter) stainless steel reactor having an agitator and several inlets for the addition of the aforementioned charges was added 61.0 parts of deionized water, 0.01 part of a 1 percent solution of the trisodium salt of N-(hydroxyethyl) ethylene diamine triacetic acid (chelant), 3.5 part of a 305 Angstrom number average diameter, 96:4 weight ratio styrene/acrylic acid copolymer latex. The reactor was purged with nitrogen, agitated at 350 rpm and heated to 90°C. Monomer feed charge (a) was added to the reactor, beginning at time $t=0$ minutes over a total period of 390 minutes. After 15 minutes, monomer charge (b) was added over a total period of 330 minutes. Charge (c) was added to the reactor beginning at time $t=0$ for a total of 330 minutes. After complete addition of all charges reaction was continued at 90°C under nitrogen atmosphere for one hour.

The resulting latex (styrene/itaconic acid = 98/2) contained 10 percent t-dodecyl mercaptan and 5 percent carbon tetrachloride remnants and had apparent number average molecular weight 1,600.

30 Sample 2

In a manner similar to Sample 1, ultra low molecular weight latex particles were prepared with a monomer composition of 78 parts styrene, 20 parts butadiene and 2 parts itaconic acid. The content of t-dodecyl mercaptan was 10 percent. Carbon tetrachloride remnant content was 5 percent.

35 Samples 3-7

Additional tackifier components were prepared by a continuous addition emulsion process similar to that disclosed in Sample 1 excepting that a mixture of styrene and butadiene monomers was charged as monomer charge (a). The amounts of monomers were varied to provide the polymer compositions indicated in Table I. Monomer charge (a) containing styrene and butadiene in total equivalent amounts equal to the equivalent of styrene used in Sample 1, along with carbon tetrachloride and t-dodecyl mercaptan, was added to the reactor, beginning at time $t=0$ minutes for a total of 270 minutes. Remaining charges (b) and (c) are as previously described. Charge (b) was added to the reactor beginning at time $t=15$ minutes for a total of 210 minutes. Charge (c) was added to the reactor beginning at time $t=0$ for a total of 300 minutes. Following addition of all feed streams, the reactor contents were agitated for an additional 60 minutes at 90°C under nitrogen atmosphere. The reactor was cooled and the resulting latex recovered.

Results of the polymerization are contained in Table I.

Table I

Sample	Feed Monomer Ratio Styrene/- Butadiene/- Itaconic Acid	Chain Transfer Agent Content (%)		Apparent Number Average Molecular Weight
		t-dodecyl mercaptan	CCl ₄	
3	78/20/2.0	10	5	2,200
4	68/30/2.0	10	5	2,400
5	58/40/2.0	10	5	2,600
6	88/10/2.0	15	5	1,330
7	88/10/2.0	20	5	1,100

Sample 8

Monomer charge (a) was prepared comprising 68 parts of styrene, 30 parts of butyl acrylate and 2 parts of t-dodecyl mercaptan.

Monomer charge (b) was prepared comprising 18.0 parts of deionized water, 2.0 parts of itaconic acid and 0.22 parts of sodium hydroxide.

Monomer charge (c) was prepared comprising 16.0 parts of deionized water, 0.18 parts of sodium hydroxide, 1.0 parts of sodium dodecyl benzene sulfonate surfactant, and 1.5 parts of sodium persulfate.

To a one gallon stainless steel reactor having an agitator and several inlets for addition of the aforementioned charges were added 77.0 parts of deionized water, 0.01 parts chelant, 10.0 parts of carbon tetrachloride and 3.5 parts of a 305 Angstrom number average diameter, 96:4 weight ratio styrene/acrylic acid copolymer seed latex. The reactor was purged with nitrogen, agitated at 350 rpm and heated to 90°C. Monomer charge (a) was added to the reactor at time t=0 for a total of 270 minutes, monomer charge (b) was added to the reactor beginning at t=0 for total of 6 minutes. Monomer charge (c) was added to the reactor beginning at time t=0 for a total of 270 minutes. Reaction was continued for one hour at 90°C following complete addition of all the charges.

Sample 9

The procedure of Sample 8 was repeated excepting that a mixture of styrene, butadiene and butyl acrylate monomers is employed as monomer charge (a). Results are contained in Table II.

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Table II

Sample	Monomer Ratio in Feed				Chain Transfer Agent Content (%)	Apparent Number Average Molecular Weight
	Styrene	Buta-diene	Butyl Acrylate	Itaconic Acid		
8	68	0	30	2	2	3,800
9	68	20	10	2	2	3,300

Manufacture of Adhesive ComponentSample A

A carboxylated styrene/butadiene latex was prepared by the continuous feed polymerization of 70 parts butadiene, 28 parts styrene, 2 parts itaconic acid, 2.25 parts t-dodecyl mercaptan, and 3.0 parts carbon tetrachloride. The resulting latex had a particle size of about 1050 Angstroms.

Sample B

A carboxylated styrene/butadiene latex was prepared essentially according to the method of Sample 10, excepting that the amount of t-dodecyl mercaptan was measured to 2.75 parts. The resulting latex had a particle size of about 1050 Angstroms.

Sample C

A carboxylated styrene/butadiene latex was prepared substantially according to the technique of Sample A, excepting that 65 parts of butadiene, 33 parts styrene, 2 parts itaconic acid, 2 parts t-dodecyl mercaptan and 3 parts of carbon tetrachloride were employed. The resulting latex had a particle size of 1050 Angstroms.

Preparation and Testing of Two Component AdhesiveExamples 1-9

Several tests were used to evaluate the adhesion of various formulations to specific substrates as well as for specific end-use applications.

(a) Adhesion to Specific Substrates: In this method, a substrate to which adhesion was to be determined was secured to a glass plate using double-sided adhesive tape. Two parallel strips of double-sided adhesive tape were then attached to the substrate approximately 6 inches (15.24 cm) apart. The adhesive was applied in an even film to the exposed surface of the substrate between the strips of tape. A reinforcing material (i.e., cheese cloth) was laid evenly over the adhesive and the strips of double-sided adhesive tape. The composite was allowed to dry for 24 hours and then conditioned overnight at a constant temperature of 70 to 72°F (21 to 22°C) and 50 percent relative humidity. One inch (2.54 cm) wide strips were cut between the strips of adhesive tape. The cheese cloth and the given substrate were separated at the double-sided tape and inserted in opposite jaws of an Instron® adhesion tester. Peel adhesion (180°) was then measured at peel rates of 2 inch/min. (5.08 cm/min.) and 20 inch/min. (50.8 cm/min.).

Several adhesive systems were formulated and evaluated by test method (a) to determine adhesion to untreated polypropylene film (Mobil Bicor® 278T). Results are contained in Table III.

Table III

Latex	Peel Adhesion, lbs/inch (kg/cm)
Sample B	1.9 (0.34)
Example 1 (Sample B/Sample 2*)	4.5 (0.81)

* blended in a 60/40 ratio

10 The results indicate that improved peel adhesion was obtained by a blend of an adhesive component and a tackifier component according to the invention.

Adhesion Test for Film To Paper Laminates:

15 This test was used to determine the performance of a given adhesive system for film to paper or paperboard laminated structures. Adhesive was applied to a 7x14 inch (17.8 cm x 35.6 cm) unbleached Alabama Kraft® paperstock (20 point). The adhesive was drawn down in the cross machine direction using a No. 31 mayer rod. After the adhesive was drawn down, the untreated side of a polypropylene film (Mobil Bicor®278T) was placed on the wet adhesive without applying pressure. The laminate was allowed to air dry for 24 hours at room temperature. Then several 1/2 inch (1.27 cm) cuts were made on the laminates in the machine direction. The laminate was attached in a horizontal position and the film edge was held between the thumb and index finger and pulled by hand in a direction about 135° to the horizontal plane. Two rates of pull, a slow rate (approximately 180 inches/minutes (457.2 cm/minutes)) and a fast rate (approximately 720 inches/minutes (1828.8 cm/minutes)) at two temperatures 40°F and 72°F (4°C and 22°C) were employed.

20 The method of bond failure was then evaluated. Fiber tear (FT) was determined by the presence of fiber particles on the separate film surface. Adhesive failure (AF) indicates clean separation of either the paperboard or the film from the adhesive layer while cohesive failure (CF) indicates separation of the adhesive layer into two or more portions.

25 The same adhesive systems which were evaluated for basic adhesion to untreated polypropylene film were evaluated by test method (b) to determine the adhesive ability to bond untreated polypropylene film to paperboard. Results are contained in Table IV.

Table IV

Adhesive System	72°F (22°C)		40°F (4°C)	
	Slow Pull	Fast Pull	Slow Pull	Fast Pull
Sample A	AF	AF	AF	AF
Example 2 (Sample B/ Sample 2*)	FT	FT	FT	FT/AF

50 * blended in a 60/40 ratio

55 Inspection of the data in the above table shows that the adhesive component by itself does not give suitable fiber tear properties. However, when such components were formulated with a tackifier component according to the invention, the resulting adhesive demonstrated that a system can effectively bond untreated polypropylene film to other substrates such as paper stock or paperboard.

(c) 180° Peel Adhesion Test for Film-Paper Laminates: Laminates were prepared as described in the above section (b). One inch (2.54 cm) wide cuts are made through the film and the paperboard. The films were pulled on an Instron® adhesion tester at 180° with a rate of pull of 12 inch/min. (30.48/min.)

at room temperature. The force required to effect destruction of the bond was recorded along with the mode of failure.

Film/Paper Adhesion

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Several adhesive systems were evaluated by test method (c) to determine adhesion between corona treated film (Oppalyte® TW-3 available from Mobil Chemicals, Inc.) and paperboard substrates. Sample portions of film were coated with various adhesive formulations. Results are contained in Table V.

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Table V

Adhesive	180° Peel Adhesion, lbs/in (kg/cm)
Sample C	480 (86.4)
Example 3 (Sample C/Sample 3; 50/50)	600 (108)
Example 4 (Sample C/Sample 4; 50/50)	900 (162)
Example 5 (Sample C/Sample 5; 50/50)	640 (115)
Example 6 (Sample C/Sample 6; 50/50)	780 (140)
Example 7 (Sample C/Sample 7; 50/50)	850 (153)
Comparative Example 1 (Sample C/PS [*] ; 50/50)	570 (103)

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*Comparative, PS is a low molecular weight polystyrene dispersion tackifier, Piccolastic A-75®, available from Hercule, Inc.

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The results of the above testing indicate that adhesion equals, or even exceeding that obtainable utilizing conventional adhesive formulation technology utilizing conventional tackifiers, can be obtained according to the present invention.

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(d) Adhesive Test for Film to Film Laminates: This method consists of applying the adhesive to primary film substrate and drawing it down the web using a No. 3 mayer rod, which is equivalent to applying approximately 0.5 to 0.7 lbs per/1000 ft² (0.22 to 0.32 kg/92.9 sq. m). This structure was then dried in a forced air oven for approximately three minutes. A secondary film substrate was applied over the coated web. A sheet of paper was placed over the laminate to protect it and an iron set at low heat (surface temperature approx. 150° F (66° C)) was run across the paper with mild hand pressure. Laminates are cut into 1 inch (2.54 cm) width strips and tested by the T-peel test (ASTM D-1876-72) at 12 inch/min. (30.48 cm/min.). The resulting force was reported in lbs (force)/inch (kg/cm).

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An adhesive system was evaluated by test method (d) to determine adhesive performance in film to film laminations. Accordingly, two corona treated polypropylene films (Mobil Bicor® 278T) were laminated by the above procedure. T-peel adhesion results are contained in Table VI.

Table VI

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Adhesive System	T-Peel Adhesive, lbs/in (kg/cm)
Sample C	200 (36)
Example 8 (Sample C/Sample 3 [*])	425 (77)

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* Blended in a 40/60 ratio

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(e) Pressure Sensitive Adhesive Test: The adhesive properties of a candidate pressure sensitive adhesive were measured according to the Pressure Sensitive Tape Council (PSTC) Procedure Nos. 1, 5 and 7. The peel adhesion test involved peeling the tape at a 180 degree angle after application under relatively heavy pressure. The quick stick test involved peeling the tape off at a 90 degree angle after application under a relatively light pressure. Both tests are reported as the force required to remove the tape. Shear adhesion was measured according PSTC No. 7 using a 500 grams (g) mass at 120° F (49° C). In this test, a shear force of 2 degree angles was applied in an oven. The results are reported as

the time required for the bond to fail at a given temperature.

Several adhesive systems were evaluated by test method (e) to determine pressure sensitive adhesive properties. Tests were conducted at 72°F (22°C). Results are contained in Table VII.

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Table VII

Adhesive Formulation	Peel Adhesion, lbs/in (<u>kg/cm</u>)	Quick Stick, lbs/in (<u>kg/cm</u>)	Shear (Hours)
Sample D	1.8 (0.3)	0.8 (0.1)	>100
Example 9 (Sample D/ Sample 3; 60/40)	3.6 (0.6)	1.2 (0.2)	80
Comparative Example II (Sample D/Conventional Tackifier*; 60/40)	3.6 (0.6)	1.4 (0.3)	>100
Comparative Example III (Sample D/Conventional Tackifier**; 60/40)	3.0 (0.5)	1.6 (0.3)	25

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*Comparative, polystyrene tackifier Piccolastic A-75, available from Hercules, Inc.

**Comparative, polystyrene tackifier, Piccolastic A-50, available from Hercules, Inc.

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The above results indicate the present invention is suitable for preparing pressure sensitive adhesives having good properties.

Claims

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1. An adhesive characterized by (1) an adhesive component comprising a random interpolymer of at least one hard monomer and (2) at least one soft monomer, having a Tg of from -70°C to -5°C, and a tackifier component that is compatible with the adhesive component, having a number average molecular weight less than 5,000 and a Tg from -10°C to 50°C, wherein both the adhesive component and tackifier component have been separately prepared by emulsion polymerization techniques.
2. An adhesive according to Claim 1, wherein the adhesive component comprises from 95 percent to 20 percent of the latex and the tackifier component comprises from 5 percent to 80 percent of the latex, such percentages being measured on a dry weight basis.
3. An adhesive according to Claim 1, wherein the hard monomer in each component is styrene and the soft monomer in each component is butadiene.

4. An adhesive according to Claim 1, wherein the tackifier component has a number average molecular weight from 1,000 to 4,000.
5. An adhesive according to Claim 1, wherein the adhesive and tackifier components comprise inter-polymers of styrene, butadiene and itaconic acid.

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REPORT

Application Number

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DOCUMENTS CONSIDERED TO BE RELEVANT		Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. Cl.8)
Category	Citation of document with indication, where appropriate, of relevant passages		
X	PATENT ABSTRACTS OF JAPAN, vol. 11, no. 89 (C-411)[2536], 19th March 1987; & JP-A-61 241 376 (IDEMITSU PETROCHEM.) 27-10-1986 "Abstract" -----	1-2	C 09 J 109/08 C 09 J 125/10 // (C 09 J 109/08 C 09 J 125:10)
TECHNICAL FIELDS SEARCHED (Int. Cl.8)			
C 09 J C 08 L			
The present search report has been drawn up for all claims			
Place of search	Date of completion of search	Examiner	
The Hague	23 April 91	SCHUELER D.H.H.	
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